

Cambridge International Examinations Cambridge International General Certificate of Secondary Education

#### **CO-ORDINATED SCIENCES**

0654/31 May/June 2017

Paper 3 Theory (Core) MARK SCHEME Maximum Mark: 120

Published

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2017 series for most Cambridge IGCSE<sup>®</sup>, Cambridge International A and AS Level and Cambridge Pre-U components, and some Cambridge O Level components.

® IGCSE is a registered trademark.

Question	Answer	Marks
1(a)(i)	cell membrane, cytoplasm and nucleus labelled ;;;	3
1(a)(ii)	any two of the following chloroplasts ; cell wall ; vacuole ;	max 2
1(b)(i)	glucose and oxygen ;	1
1(b)(ii)	any two of the following muscle contraction ; protein synthesis ; cell division ; growth ; the passage of nerve impulses ; maintenance of a constant body temperature ;	max 2

Question	Answer	Marks
2(a)(i)	conduction ;	1
2(a)(ii)	to enable / facilitate convection ; description of convection ;	2
2(a)(iii)	good (thermal) insulator ;	1
2(b)(i)	lost as heat / lost to surroundings / lost as sound energy ;	1
2(b)(ii)	useful energy out ÷ total energy in ;	1
2(c)	bare copper wire / damaged insulation ; can cause short circuit / electrocution / fire ;	2
2(d)(i)	temperature at which (all of) a liquid turns to gas ;	1
2(d)(ii)	B – particles close together and randomly arranged ; C – particles widely spaced (and randomly arranged) ;	2

Question	Answer	Marks
3(a)	78;21;	2
3(b)(i)	incomplete combustion of fuel ;	1
3(b)(ii)	toxic to humans ;	1
3(c)	green to orange / red ; solution is acidic / non-metal oxides are acidic ;	2
3(d)(i)	NH <sub>3</sub> ;	1
3(d)(ii)	elements contain only one type of atom ; compounds contain different atoms (bonded) ; any correct reference to the example molecules ;	max 2

Question	Answer	Marks
4(a)(i)	ovary where fetus develops	3
	vagina carries egg to the uterus	
	oviduct produces egg cells	
	uterus receives penis during intercourse	
	···· 733	
4(a)(ii)	oviduct ;	1
4(a)(iii)	fusion / joining of, sperm and egg ; nuclei ;	2
4(b)	only one parent ; no variation ; does not involve gametes ; less time / energy spent looking for a mate ;	max 2

May/June 201	(
--------------	---

Question	Answer	Marks
5(a)	fat butter ; protein fish ; vitamin C melon / tomatoes ;	3
5(b)(i)	growth / repair ;	1
5(b)(ii)	carbon hydrogen oxygen nitrogen ;	1
5(b)(iii)	amino acids ;	1
5(c)(i)	(teenager) more active ; has a higher metabolic rate ; is still growing ;	max 1
5(c)(ii)	(female athlete) lower metabolic rate ; generally, smaller in size / mass ;	max 1

May/June 2017
---------------

Question	Answer	Marks
6(a)(i)	H = H = H = H = H = H = H = H = H = H =	2
6(a)(ii)	all correct single bonds ; carbon dioxide ; water ;	2
6(b)(i)	fractional distillation;	1
6(b)(ii)	similarities hydrocarbons / mixtures / contain alkanes / other correct ; differences boiling point/range / colour / odour / flammability / other correct ;	2
6(c)(i)	bromine ;	1
6(c)(ii)	no change / colour remains / solution stays orange ; bromine does not react with saturated hydrocarbons ;	2
6(c)(iii)	ethene ;	1

Question	Answer	Marks
7(a)(i)	when time is 0 s / 40–50 s ;	1
7(a)(ii)	4 (m / s) ;	1
7(a)(iii)	distance = speed $\times$ time / 5 $\times$ 4 ; = 20 (m) ;	2
7(b)	reference to friction or description / transfer of electrons / negative charge ;	1
7(c)	no deviation at first interface and first reflection correct ; second reflection correct ;	2
7(d)(i)	move faster ;	1
7(d)(ii)	more frequent collisions / collide at greater speed (with wall) ; more force exerted on tyre walls ;	2

Question	Answer	Marks
8(a)	one arrow pointing into the roots ; one arrow pointing out from the leaf ;	2
8(b)	xylem ;	1
8(c)	light / carbon dioxide / chlorophyll ;	1
8(d)	higher temperature / hot ; increased wind speed / windy ; arid / dry (conditions) ;	max 2

Question	Answer	Marks
9(a)(i)	12 ; 14 ; 12 ;	3
9(a)(ii)	8;	1
9(b)(i)	hydrogen ;	1
9(b)(ii)	increases ; mixture becoming less acidic / the acid is being used up / is becoming less ;	2
9(b)(iii)	(incorrect) reaction is exothermic ; because temperature increased / endo thermic would show temperature decrease ; reference to transfer of chemical to thermal energy / or vv if endothermic ;	max 2
9(c)	increase acid concentration ; increase temperature ; increase surface area of magnesium ;	max 2

Question	Answer	Marks
10(a)	flow of energy ; from one organism to the next ;	2
10(b)(i)	grass $\rightarrow$ rabbit $\rightarrow$ fox / grass $\rightarrow$ rabbit $\rightarrow$ hawk ;;	2
10(b)(ii)	producer grass / blackberries ; carnivore hawk / fox ; herbivore butterfly / grasshopper / mouse / rabbit ;	3
10(c)	fewer, mice / rabbits eaten by hawks ; less <u>competition</u> ; more, food / mice / rabbits (for foxes) ; fox population increases ;	max 3

Question	Answer	Marks
11(a)	chemical ; water ; turbine ;	3
11(b)(i)	use a geiger counter ; ref to passing through lead etc. ;	2
11(b)(ii)	cancer etc. ;	1
11(c)(i)	resistance increases ;	1
11(c)(ii)	change length / material ;	1
11(d)	contract in cold weather ; damage cables / pylons ;	2

Question	Answer	Marks
12(a)(i)	<b>S</b> ; insulator (shows it is a non-metal) / liquid because m.pt. less than RT and b.pt. greater than RT ;	2
12(a)(ii)	high density / malleable / sonorous / lustrous / conducts heat (well) ;	1
12(b)(i)	sodium chloride ;	1
12(b)(ii)	loses electrons / an electron ;	1
12(b)(iii)	opposite charges (attract) ;	1
12(c)(i)	electrolysis ;	1
12(c)(ii)	Y anode ; Z cathode ;	2

Question	Answer	Marks
13(a)	gravitational (potential) energy ;	1
13(b)	measure the number of seconds / time between noise and echo ; divide distance by time ; divide double the distance / multiply by 2 ;	3
13(c)	infra-red to right of visible ; ultraviolet to left of visible ;	2
13(d)(i)	middle ray passes through without deviation and bottom ray passes out parallel to principal axis ;	1
13(d)(ii)	inverted arrow drawn at intersection of three rays ;	1
13(d)(iii)	principal focus / focal point ;	1